Searching for the Great Oxidation Event in North America: A Reappraisal of the Huronian Supergroup by SIMS Sulfur Four-Isotope Analysis

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Abstract

Sedimentological observations from the Paleoproterozoic Huronian Supergroup are suggested to mark the rise in atmospheric oxygen at that time, which is commonly known as the Great Oxidation Event (GOE) and typically coupled with a transition from mass-independent fractionation (MIF) to mass-dependent fractionation (MDF) of sulfur isotopes. An early in situ study of S three-isotopes across the Huronian Supergroup by Papineau et al. (2007) identified a weak MIF–MDF transition. However, the interpretation and stratigraphic placement of this transition is ambiguous. In this study, all four S isotopes were analyzed for the first time in two Huronian drill cores by secondary ion mass spectrometer (SIMS), and both $\Delta^{33}S$ and $\Delta^{36}S$ were calculated. Based on improved precision and detailed petrography, we reinterpret the dominance of pyrrhotite in the studied sections, which was previously proposed as “early authigenic” in origin, as resulting from regional metamorphism. Small but analytically resolvable nonzero values of $\Delta^{33}S$ (from $-0.07\%$ to $+0.38\%$) and $\Delta^{36}S$ (from $-4.1\%$ to $+1.0\%$) persist throughout the lower Huronian Supergroup. Neither pronounced MIF-S signals nor a MIF–MDF transition are seen in this study. Four scenarios are proposed for the genesis of small nonzero $\Delta^{33}S$ and $\Delta^{36}S$ values in the Huronian: homogenization by regional metamorphism, recycling from older pyrite, dilution by magmatic fluids, and the occurrence of MDF. We argue that the precise location of the MIF–MDF transition in the Huronian remains unsolved. This putative transition may have been erased by postdepositional processes in the lower Huronian Supergroup, or may be located in the upper Huronian Supergroup. Our study highlights the importance of integrated scanning electron microscopy and secondary ion mass spectrometry techniques in deep-time studies and suggests that different analytical methods (bulk vs. SIMS) and diagenetic history (primary vs. metamorphic) among different basins may have caused inconsistent interpretations of S isotope profiles of the GOE successions at a global scale. Key Words: Great Oxidation Event (GOE)—Secondary ion mass spectrometer (SIMS)—Paleoproterozoic—Sulfur isotopes—Mass independent fractionation (MIF). Astrobiology 18, 519–538.

1. Introduction

A unique feature of Earth is the existence of free oxygen in the atmosphere. However, early Earth largely lacked atmospheric oxygen for its first ~2.2 billion years (Cloud, 1972; Holland, 1978; Holland et al., 1986; Farquhar et al., 2011, 2014; Olson et al., 2018). It is widely regarded that the rise of atmospheric oxygen on Earth has profoundly influenced the evolutionary trajectory of the atmosphere (Holland, 1978), the hydrosphere (Lyons et al., 2014; Liu et al., 2016), the lithosphere (Hazen et al., 2008; Hazen and Ferry, 2010), and life on this planet (Cloud, 1968, 1976; Xiao, 2014; Knoll, 2015; Knoll et al., 2016).

Sedimentological observations worldwide suggest that Earth witnessed the first rise in atmospheric oxygen at ~2.3 Ga, commonly known as the Great Oxidation Event (GOE) (Holland, 1984; Kasting et al., 1992; Bekker et al., 2004; Hannah et al., 2004; Holland, 2006). This event has been widely inferred by a global sedimentological transition from the disappearance of oxygen-sensitive detrital minerals (e.g., pyrite, uraninite, and siderite) and the widespread appearance of hematite-rich red beds in the sedimentary record.

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Cloud, 1968; Roscoe, 1969, 1973; Rasmussen and Buick, 1999; Dutkiewicz et al., 2006; Farquhar et al., 2011; Johnson et al., 2014; Fischer et al., 2016). Coincidentally, geochemical constraints based on sulfur three-isotopes of sedimentary authigenic pyrite reveal notable disappearance of mass-independent fractionation (MIF-S) in the Paleoproterozoic, followed by over 2 billion years characterized by mass-dependent fractionation (MDF-S) (Farquhar et al., 2000; Farquhar and Wing, 2003, 2005; Johnston, 2011; Ono, 2017). The discovery of the MIF–MDF transition is widely regarded as the smoking gun for the GOE (Kump, 2008).

The Huronian Supergroup in Canada (Figs. 1 and 2) is one of the most-studied Paleoproterozoic sedimentary records in the world and thus bears the potential to capture the GOE. Intensive studies have been conducted on the sedimentology (e.g., Wood, 1973; Young, 1991; Hill et al., 2016), geochemistry (e.g., Krogh et al., 1984; Corfu and Andrews, 1986; Ketchum et al., 2013; Rasmussen et al., 2013), paleomagnetism (e.g., Williams and Schmidt, 1997), biomarkers (e.g., Dutkiewicz et al., 2006), and chemosтратigraphy (e.g., Nesbitt and Young, 1982; Wing et al., 2002, 2004; Bekker and Kaufman, 2007; Papineau et al., 2007; Sekine et al., 2011a, 2011b; Goto et al., 2013) of the Huronian Supergroup.

Notably, a sulfur three-isotope study by secondary ion mass spectrometer (SIMS) was published showing mild MIF(\(\Delta^{33}S\)) signals in rocks deposited below the Bruce diamictite (McKim and Pecors Formations) and MDF(\(\Delta^{33}S\)) signals above the Bruce diamictite (in the Espanola Formation) (Papineau et al., 2007). This tantalizing but subtle shift in the Huronian Supergroup has been used to correlate and infer the global patterns of the GOE among other Paleoproterozoic successions worldwide (e.g., Papineau et al., 2007; Williford et al., 2011; Hoffman, 2013; Rasmussen et al., 2013; Gumsley et al., 2017).

However, the published \(\Delta^{33}S\) data from the Huronian Supergroup all fall within a narrow range of \(\Delta^{33}S = -0.25\%_o \text{ to } +0.25\%_o\) (ca. \(\pm 0.23\%_o\) precision reported as 2SD hereafter), except three data points (0.45\%_o, 0.51\%_o, and 0.88\%_o) that have been interpreted as critical MIF-S signals marking the end of the GOE (Papineau et al., 2007). The absence of a stronger MIF–MDF transition in the Huronian stands in strong contrast to other GOE sections where \(\Delta^{33}S\) values are up to \(+7\%_o\) (Guo et al., 2009; Luo et al., 2016) and suggested the need for a more comprehensive study including \(\Delta^{36}S\) analysis.

Questions based on the published \(\Delta^{33}S\) record in Huronian sediments include the interpretation of faint MIF and MDF signals and thereby the location of MIF–MDF transition and of the GOE. (1) Might the small nonzero \(\Delta^{33}S\) values be an artifact of poorer analytical precision in early studies? (2) If real, are the small nonzero \(\Delta^{33}S\) values due to MIF? Alternatively, these small MIF signals might reflect remobilized compositions from an older MIF-bearing sulfur reservoir (e.g., Ulrich et al., 2011; Williford et al., 2011; Cabral et al., 2013; Reinhard et al., 2013), a period with diminished photochemical reaction in the atmospheric S cycle (Williford et al., 2011), or some other conditions that favor the production of a prolonged, muted \(\Delta^{33}S\) signature (e.g., Thomazo et al., 2009). (3) Does the weak MIF–MDF transition in the lower Huronian (Papineau et al., 2007) reflect variation in secular S isotope signals or values altered by regional metamorphism? It should be taken as cautionary.

FIG. 1. Map showing the distribution of the Huronian Supergroup and other Paleoproterozoic successions in the Great Lakes area. The two studied drill cores (150–4, 156–1) are located ca. 10 km north of the city of Elliot Lake, Ontario, Canada. Modified from the work of Rasmussen et al. (2013).
FIG. 2. Integrated lithostratigraphy, sedimentology, and geochemical studies of the Huronian Supergroup. Lithostratigraphy modified after Young et al. (2001), Dukiewicz et al. (2006), and Hill et al. (2016). Sedimentological observations include detrital pyrite and uraninite (Roscoe, 1969), oil-bearing fluid inclusions (Dukiewicz et al., 2006), red beds (Roscoe, 1973), microbially induced sedimentary structures (Hill et al., 2016), and sulfate nodules (Chandler, 1988). Age constraints were measured from Nipissing diabase (Corfu and Andrews, 1986) and Copper Cliff Formation (Krogh et al., 1984; Ketchum et al., 2013). Geochemical data include sulfur isotopes (Hattori et al., 1983; Papineau et al., 2007), CIA (Nesbitt and Young, 1982), high initial $^{187}$Os/$^{188}$Os, RSE (Sekine et al., 2011a), and MnO–Fe$_2$O$_3$ enrichment right after the Gowganda glaciation (Sekine et al., 2011b). Note that the previously proposed MIF–MDF transition (Papineau et al., 2007) has been reevaluated in this study. Abbreviations: CIA = chemical index of alteration; MDF = mass-dependent fractionation; MIF = mass-independent fractionation; MISS = microbially induced sedimentary structure; RSE = redox-sensitive elements.
that most published Huronian $\Delta^{33}S$ data points can hardly be 
distinguished between MIF and MDF.

To answer these questions, we conducted a SEM–SIMS 
study that combines petrography and imaging with in situ 
S four-isotope study of the Huronian sulfide minerals by 
SIMS. Improved analytical precision of all four S isotopes 
was achieved with modifications of the SIMS detector sys-
tem and analytical protocols (Ushikubo et al., 2014). Both 
$\Delta^{33}S$ and $\Delta^{34}S$ were determined for the first time by SIMS in 
the Huronian. Two of the previously studied drill cores of 
the Huronian Supergroup were analyzed to reevaluate the 
dynamics of biogeochemical S cycles and atmospheric ox-
ygen evolution in this critical period.

2. Geological Background

The Huronian Supergroup is comprised of five groups: 
Elliot Lake, Hough Lake, Quirke Lake, Cobalt, and Flack 
Lake, in ascending stratigraphic order (Fig. 2). There are 
three distinct glacial diamictite deposits: Ramsay Lake, 
Lake, in ascending stratigraphic order (Fig. 2). There are 
Elliot Lake, Hough Lake, Quirke Lake, Cobalt, and Flack 
Lake, in ascending stratigraphic order (Fig. 2). There are 
Elliot Lake, Hough Lake, Quirke Lake, Cobalt, and Flack 
Lake, in ascending stratigraphic order (Fig. 2). There are

The upper Huronian Supergroup starts with the glacial 
Gowganda diamictites. Above the uppermost Gowganda 
diamictite, hematite-rich red beds are found in the Lorrain 
Formation, which includes radiometric ages of zircons from the felsic volcanic 
rocks in the Copper Cliff Formation near the base of the 
supergroup (ca. 2.45 Ga) (Kroghe et al., 1984; Ketchum et al., 
2013) and U–Pb zircon ages (ca. 2.3 Ga) in the Gordon Lake 
Formation (Rasmussen et al., 2013). Concordant bandedite 
and rutile ages from the Nipissing diabase dike swarm at ca. 2.22 Ga (Mossman et al., 1993).

3. Materials and Methods

3.1. Sampling strategy

In this study, to help avoid detrital components, only ma-
rine facies (i.e., carbonate- or silstone-dominated intervals) 
were selected for petrographic and geochemical analysis. The 
sampled intervals include the McKim Formation (marine 
silstone), the Pecors Formation (marine silstone), and the 
Espanola Formation (marine silstone and limestone). Due to 
the large contribution by detrital/terrestrial sediments in silici-
clastic facies (e.g., Ulrich et al., 2011; Zhou et al., 2017), 
the sandstone-dominated Mississagi and Serpent formations 
and the three diamictite intervals were avoided during sam-
plying. A suite of 16 samples from two Huronian drill cores 
was analyzed for S four-isotopes by SIMS, including 11 
samples from the Kerr-McGee drill core 150–4 (Supple-
ment S2; Supplementary Data are available at http:// 
online.liebertpub.com/doi/suppl/10.1089/ast.2017.1722) and 
5 samples from the Kerr-McGee drill core 156–1 (Supple-
ment S3). Based on the Ontario Geological Survey map #2419, 
these two cores are located ca. 10 km north of the city of 
Elliot Lake, Ontario, Canada (Fig. 1; Fig. 1 in Papineau et al., 
2007). All the drill core samples are stored in the Ontario 

Several sulfide minerals have been found in the Huron-
ian samples, which include pyrite, pyrrhotite, chalcopyrite, 
 sphalerite, pentlandite, and cobaltite. Due to “orientation 
effects” that degrade accuracy for SIMS data in a few minerals 
(Kozdon et al., 2010; Kita et al., 2011), sphalerite was not 
considered for analysis in this study. Only grains of pyrite, 
pyrrhotite, and chalcopyrite, that have been shown not to have 
orientation effects, were analyzed by SIMS for S four-isotopes 
in this study. Most of the sulfide minerals are anhedral in 
shape, showing irregular morphology commonly inter-
growing with chlorite and other silicate minerals (Fig. 3). 
SIMS samples were cast in round mounts with the UWPy-1 
pyrite standard near the center of the 25 mm diameter, 5 mm 
thick mount. All analyzed minerals were imaged by scan-
nng electron microscope (SEM) with back-scattered elec-
trons (BSE) prior to SIMS analysis. Mineral chemistry was 
verified by SEM energy-dispersive spectrometry (EDS) and 
wave-length-dispersive electron-probe microanalysis (EPMA).

3.2. SIMS analysis

Sulfur four-isotope data were measured with a CAMECA 
IMS 1280 in the WiscSIMS (Wisconsin Secondary Ion Mass 
Spectrometer) Lab at the University of Wisconsin–Madison. 
Detailed methods can be found in previous publications 
(Valley and Kita, 2009; Kozdon et al., 2010; Williford et al., 
2011, 2016; Ushikubo et al., 2014) and are only briefly 
described below.

Analysis of sulfur four-isotopes employed a $^{133}$Cs$^+$ beam 
($\sim$ 5 nA) accelerated at 10 kV (impact energy = 20 kV). The 
beam size in this study is 20 $\mu$m. Samples were coated with
60 nm of gold, and a normal-incidence electron flood gun was used for charge compensation. Secondary ions of $^{32}$S-, $^{33}$S-, $^{34}$S-, and $^{36}$S- were detected simultaneously using four Faraday cup detectors with modified slits as described by Ushikubo et al. (2014). UWPy-1 (pyrite) was used as running standard with $\delta_{\text{CDT}}^{34}$S = 16.04 ± 0.18‰, $\Delta^{33}$S = -0.003 ± 0.009‰, and $\Delta^{36}$S = -0.21 ± 0.24‰, values analyzed by conventional gas source mass spectrometer (Ushikubo et al., 2014; Williford et al., 2016). The running standard is mounted in the center of each SIMS sample mount together with the unknown Huronian samples in order to closely monitor the running conditions during SIMS analysis. The raw isotope ratios were corrected for instrumental bias using mineral-specific lab standards: UWPy-1, Anderson pyrrhotite, and Trout Lake chalcopryite (Crowe and Vaughan, 1996; Kozdon et al., 2010) (Supplement S1). Analytical spot-to-spot precisions are $\delta^{34}$S ± 0.39‰, $\Delta^{33}$S ± 0.04‰, and $\Delta^{36}$S ± 1.13‰ (2SD) based on the groups of eight bracketing UWPy-1 standard analyses during the SIMS session of Huronian samples (Supplement S4), which is comparable to what Ushikubo et al. (2014) reported in previous sessions ($\delta^{34}$S ± 0.23‰, $\Delta^{33}$S ± 0.05‰, $\Delta^{36}$S ± 0.86‰).

UWPy-1 was analyzed a total of 157 times in this study (Supplement S4).

**FIG. 3.** Back-scattered electron (BSE) images of the Huronian samples show close association between pyrrhotite (Po) and chlorite (Cht). Images B, D, F, H show the marked zones in images A, C, E, G, respectively. Sample depth: (A–D) 4625.1 m in drill core 156–1; (E, F) 4559 m in drill core 156–1; (G, H) 4625.1 m in drill core 156–1; (I–L) 5219 m in drill core 150–4.
Each analysis consisted of 10 s for pre-sputtering, 80 s for centering of secondary ions in the field aperture, and 80 s for isotope measurement. Average count rates for $^{32}\text{S}$ confirmed the stability of individual analyses. A typical count rate of $^{32}\text{S}$ was $\sim 5 \times 10^9$ counts per second for pyrrhotite and pyrite, and $\sim 6 \times 10^9$ counts per second for chalcopyrite. To monitor the magnitude of interference from $^{32}\text{SH}$ on the $^{33}\text{S}$ signal, the $^{32}\text{SH}$ peak was measured by scanning the deflector located between the magnet and the Faraday cup detectors after each analysis (Williford et al., 2011). The ratio of the $^{32}\text{SH}$ tail at the $^{33}\text{S}$ peak position relative to the $^{33}\text{S}$ peak ($^{32}\text{SH}_{\text{tail}}/^{33}\text{S}_{\text{peak}}$) was determined to be $\sim 1 \times 10^{-3}$, which was used to correct the contribution of the $^{32}\text{SH}$ tail signal to the $^{33}\text{S}$ peak. The contribution of $^{32}\text{SH}$ to the $^{33}\text{S}$ in all the spots is negligible (with corrections from 0.0001 to 0.001).

After SIMS analysis, gold coating of the samples was removed by chemical dissolution of gold with saturated aqueous solution of potassium iodide (Jones et al., 2012). Samples were recoated with 20 nm thickness of carbon, and each pit was investigated by SEM-SE for possible irregularities.

### 3.3. Sulfur four-isotope data correction

Sulfur isotope ratios are reported in standard per mil ($^{\circ}$) notation relative to Vienna Canyon Diablo Troilite (VCDT). Measured ratios of $^{33}\text{S}/^{32}\text{S}$, $^{34}\text{S}/^{32}\text{S}$, and $^{36}\text{S}/^{32}\text{S}$ were divided by VCDT values and calculated as “raw” $\delta$-values: $\delta^{34}\text{S}_{\text{raw}}$, $\delta^{33}\text{S}_{\text{raw}}$, and $\delta^{36}\text{S}_{\text{raw}}$, respectively. The sulfur isotope ratios of VCDT used in the calculations are $^{34}\text{S}/^{32}\text{S}=1/126.948$ (Ding et al., 2001), and a revised $^{36}\text{S}/^{32}\text{S}$ value of 1/6641 (Ushikubo et al., 2014) instead of 1/6515 (Ding et al., 2001). Correction for $^{32}\text{SH}$ contribution to the $^{33}\text{S}$ peak was applied at this step. In this study, UWPY-1 (Balmat pyrite) was used as the monitoring standard for all the samples.

Correction factors of the instrumental bias of $^{34}\text{S}$ for pyrite samples are calculated based on the S four-isotope values of UWPY-1 measured by conversion of powdered sample to $\text{SF}_6$ and gas-source mass spectrometry (Ushikubo et al., 2014). Although only the UWPY-1 pyrite standard was mounted in the center of each SIMS sample, pyrrhotite and chalcopyrite standards in other standard mounts were also analyzed at the beginning of each session in order to constrain the instrumental bias between different sulfide minerals (Supplement S1). In addition, Rutan pyrite and Norilsk chalcopyrite were run as checks for accuracy. Relative bias between pyrrhotite and pyrite was calculated based on the offset between UWPY-1 pyrite standard and Anderson pyrrhotite standard. All the corrected data, together with the corresponding petrographic images by SEM, can be found in the online supplementary materials (Supplements S1–S4).

### 3.4. EPMA analysis

Major element concentrations of sulfide minerals, including pyrrhotite, chalcopyrite, sphalerite, pentlandite, pyrite, and cobaltite, were analyzed by EPMA for Fe, S, Co, Ni, Cu, Zn, and As concentrations using wavelength-dispersive crystal spectrometers. Before EPMA, each spot was investigated by SEM, BSE, and EDS. EPMA was performed with a Cameca SXFive FE electron microprobe at 20 kV, 20 nA and 1 micron defocused beam, with Astemix sulfide as well as pure metal standards, using the Armstrong/Love-Scott matrix correction, with Probe for Windows software (Donovan et al., 2018) at the Eugene Cameron Electron Microprobe Lab, Department of Geoscience, University of Wisconsin–Madison.

### 4. Results

#### 4.1. Petrographic observations

Detailed petrographic investigation reveals multiple sulfide minerals, including pyrite, pyrrhotite, chalcopyrite, pentlandite, sphalerite, and cobaltite in the studied samples (Figs. 3–8; see Supplements S2–S4 for a complete log of isotope sample and standard analyses). EPMA data (Supplement S5) confirm the identification of these minerals.

Compared with other roughly time-equivalent strata that have been studied for S three- or four-isotopes (Guo et al., 2009; Williford et al., 2011; Luo et al., 2016), a distinct feature of the Huronian samples is the dominance of pyrrhotite as the main sulfide mineral. Both granular and elongated grains of pyrrhotite and subordinate pyrite were imaged and analyzed for multiple S isotopes. Elongate grains are typically aligned parallel to foliation and intergrown with chlorite (Fig. 3). Most of the pyrrhotite grains have fine pentlandite intergrowths or exsolution (Fig. 6). Cobaltite is also seen in a few samples; grains are mostly euhedral and cutting across the pyrrhotite (Fig. 8), suggesting a relatively late-stage mineralization.

#### 4.2. SIMS S four-isotope results

Chemostratigraphic profiles of the Huronian Supergroup in Fig. 9 present all the new data measured by SIMS in this study. The $\delta^{33}\text{S}$ profiles of both drill cores show mostly positive values from 0 to +5, except for a few slightly negative values in the McKim Formation of drill core 150–4 (Fig. 9A), and notably positive $\delta^{34}\text{S}$ data of ca. +15 in chalcopyrite of the Espanola Formation of the drill core 156–1 (Fig. 9D). In drill core 150–4, $\Delta^{33}\text{S}$ values of this study range from +0.004 to +0.38 (Fig. 9B), while in drill core 156–1, the $\Delta^{33}\text{S}$ data range from -0.07 to +0.26 (Fig. 9E) (average 2SD = ±0.04 for an individual analysis). Compared with the earlier study (Papineau et al., 2007), we also simultaneously analyzed $\Delta^{34}\text{S}$. New $\Delta^{34}\text{S}$ values analyzed from the same Huronian samples range from -4.1 to +1.0 in drill core 150–4 (Fig. 9C) and from -3.6 to +0.5 in drill core 156–1 (Fig. 9F) (average 2SD = ±1.13 for an individual analysis).

Cross-plots of $\delta^{34}\text{S}$ versus $\Delta^{33}\text{S}$ and $\Delta^{33}\text{S}$ versus $\Delta^{34}\text{S}$ are tightly clustered and do not show clear correlations or $\Delta^{33}\text{S}$ versus $\Delta^{34}\text{S}$ slopes (Fig. 10). Values of $\delta^{34}\text{S}$ from pyrrhotite and chalcopyrite are generally higher than pyrite in both drill cores 150–4 and 156–1. These $\delta^{34}\text{S}$ fractionations are reversed and do not represent equilibration during deposition or metamorphism (Ohmoto and Rye, 1979; Ohmoto, 1986; Seal, 2006). Chalcopyrite grains reveal the most positive $\delta^{34}\text{S}$ values among all sulfides.

To assess the published and new $\Delta^{33}\text{S}$ data sets, a close evaluation of the point-to-point precision is necessary. It should be noted that the published uncertainties for the $\Delta^{33}\text{S}$ data by Papineau et al. (2005) are an integration of both internal error and external error calculated using a York-based regression technique (Papineau et al., 2005, 2016).
while the reported 2SD of SIMS data in our study is the measured reproducibility of the bracketing UWPy-1 standard that were measured throughout our analysis sessions (\(n = 157\)) (Supplement S4). Thus a direct comparison of the published precisions of these two data sets may be misleading. Therefore, we have made a direct comparison of the reproducibility of analyzed sulfur isotope data for homogeneous standard material (Balmat pyrite, UWPy-1) in both data sets. The 2SD values calculated from repeated analyses of the UWPy-1 standard (\(n = 157\)) in this study are ±0.39‰ (\(\delta^{34}S\)), ±0.04‰ (\(\Delta^{33}S\)), and ±1.13‰ (\(\Delta^{36}S\)). The 2SD values of \(\delta^{34}S\) and \(\Delta^{33}S\) data are systematically 2–3 times smaller than the ones calculated from a Balmat pyrite standard (not UWPy-1, \(n = 11\)) in the work of Papineau et al. (2005) that show precisions of ±0.69‰ (\(\delta^{34}S\)), ±0.14‰ (\(\Delta^{33}S\)).

5. Discussion
5.1. Origin of the pyrrhotite in the Huronian

A distinct feature of the studied intervals in the Huronian Supergroup is the dominance of pyrrhotite among all sulfide minerals. This is in strong contrast with other roughly equivalent Paleoproterozoic successions with pyrite as the
dominant sulfide mineral (e.g., Guo et al., 2009; Williford et al., 2011; Luo et al., 2016). Therefore, interpretation of S isotope profiles of the Huronian Supergroup requires understanding on the genesis of the Huronian pyrrhotite. Pyrrhotite has been reported from both ancient strata (Reynolds et al., 1990; Reuschel et al., 2012; Asael et al., 2013; Minguez et al., 2016) and modern marine sediments (Horng and Roberts, 2006; Larrasoaña et al., 2007; Kars and Kodama, 2015; Honsho et al., 2016). However, pyrrhotite is rare in modern marine sediments and is often detrital in origin resulting from rapid weathering and short distances of transport (Horng and Roberts, 2006). In other cases, pyrrhotite has been found associated with gas hydrates (Kars and Kodama, 2015) or hydrocarbon seepage (Reynolds et al., 1990). Pyrrhotite has also been reported in ancient sedimentary records such as the ca. 2.0 Ga Pilgujärvi Sedimentary Formation in northwest Russia (Reuschel et al., 2012) and ca. 2.0 Ga Zaonega Formation in Karelia (Asael et al., 2013), which have both been interpreted to result from the decomposition of pyrite during low-grade metamorphism.

Previously, pyrrhotite in the Huronian was interpreted as early authigenic (Papineau et al., 2007). Based on this interpretation, the measured S isotope values from these pyrrhotites have been used to infer secular variation in atmospheric oxygen and marine sulfur cycle (Papineau et al., 2007) and global correlations of the GOE-bearing sections (Hoffman, 2013; Gumsley et al., 2017).

However, in this study, multiple lines of evidence suggest that pyrrhotite in the studied samples actually formed during regional metamorphism. The evidence includes

1. Irregular/anhedral shape and large grain size. Most of the pyrrhotite grains show irregular/anhedral shapes, remarkably large (up to several hundred micrometers) in size (Figs. 3–8; Supplements S2 and S3), which look unlikely to be formed in a water column or shallow marine sediment.

FIG. 5. Back-scattered electron (BSE) images with SIMS sulfur isotope data of representative pyrite from the Kerr–McGee drill cores 150–4 and 156–1 of the Huronian Supergroup. Silicate matrix is black. Abbreviations: Po = pyrrhotite; Py = pyrite; Sp = sphalerite. The upper right of each figure shows precisions of $\delta^{34}$S, $\Delta^{33}$S, and $\Delta^{36}$S with two standard deviations. All data can be found in the online supplementary material.
FIG. 6. Back-scattered electron (BSE) images with SIMS sulfur isotope data of representative pyrrhotite from the Kerr–McGee drill cores 150–4 and 156–1 of the Huronian Supergroup. Silicate matrix is black. Abbreviations: Pn = pentlandite; Po = pyrrhotite. The upper right of each figure shows precisions of $\delta^{34}S$, $\Delta^{33}S$, and $\Delta^{35}S$ with two standard deviations. All data can be found in the online supplementary material.
(2) Diverse mineral assemblages. We found that many pyrrhotite grains are parallel with metamorphic foliation and intimately associated with chlorite, pentlandite, and chalcopyrite (Figs. 3 and 4; Supplements S2 and S3), which suggest metamorphism.

(3) Homogeneous compositions. All the studied pyrrhotites have remarkably homogeneous compositions of sulfur isotope ratios and elemental abundances at micrometer scales (Figs. 5–9; Supplements S2 and S3), which suggest metamorphism.

5.2. Reevaluating the “MIF–MDF transition” in the Huronian

The sedimentological transition in the Huronian reveals the existence of detrital pyrite and uraninite in the lower Huronian and of red beds in the upper Huronian, which has been widely accepted to bracket the rise in atmospheric oxygen (Roscoe, 1969, 1973). If correct, one would expect to see a corresponding MIF–MDF transition in both $\Delta^{33}S$ and $\Delta^{36}S$ profiles of the Huronian Supergroup. The MIF–MDF transition has been proposed based on small MIF($\Delta^{33}S$) signals that were interpreted to exist in strata below the Bruce diamictite versus MDF($\Delta^{33}S$) signals in the Espanola carbonates above the Bruce diamictite (Papineau et al., 2007) (Fig. 11D, 11H).

However, the new $\Delta^{33}S$ data with improved precision suggest a different view. Our $\Delta^{33}S$ data all fall within a very
narrow range. Small nonzero $\Delta^{33}S$ values persist throughout the lower Huronian including the Espanola. Neither pronounced MIF-S signals nor a MIF–MDF transition are seen in this study (Fig. 11C, 11G).

The newly analyzed $\Delta^{36}S$ data further show nonzero signals in the lower Huronian. In contrast with the proposed “MDF-S” interval for the Espanola Formation that is based on $\Delta^{33}S$ data (Papineau et al., 2007), the $\Delta^{36}S$ data analyzed from the same Huronian samples range from $-4.1\%_{oo}$ to $+1.0\%_{oo}$ in drill core 150–4 (Fig. 9C) and from $-3.6\%_{oo}$ to $+0.5\%_{oo}$ in drill core 156–1 (Fig. 9F), which show that nonzero $\Delta^{36}S$ signals also persist throughout the lower Huronian, including the Espanola.

In summary, although the entire studied interval in the lower Huronian shows a narrow range, the nonzero $\Delta^{33}S$ and $\Delta^{36}S$ signals are still distinguishable. These small but analytically resolvable nonzero signals exist before and after both glaciations of the lower Huronian Supergroup and therefore cannot be used to locate the GOE.

5.3. Origin of the small nonzero $\Delta^{33}S$ and $\Delta^{36}S$ signals in the Huronian

Multiple hypotheses may be explained to explain the measured S isotope values in the Huronian pyrrhotites. Below, we explore these possibilities.

5.3.1. Homogenization of MIF signals by regional metamorphism. It is likely that the pyrrhotite in the studied drill cores formed by reaction of sedimentary pyrite during low-grade metamorphism. Theoretically, two general processes can lead to the transformation from pyrite (FeS$_2$) to pyrrhotite (Fe$_{0.8–1}$S): (1) gaining iron via reactions with Fe oxide or Fe-bearing silicates or (2) losing sulfur via metamorphic breakdown of pyrite into hydrogen sulfide or sulfur dioxide (Reuschel et al., 2012; Asael et al., 2013).

In the first scenario, although very few Fe oxide minerals are found surrounding the sulfide minerals (Fig. 4B, 4C) and are probably late alteration in origin, it is unlikely that iron...
Oxide is a sufficient source of iron given the rarity of Fe oxide minerals in the studied interval (below the red beds in the Lorrain Formation). However, chlorite exists in close association with pyrrhotite (Fig. 3) and may have been a source of iron during pyrite–pyrrhotite transformation.

In the second scenario, metamorphic desulfidation reactions of pyrite can evolve either \( \text{SO}_2 \) (via reaction \( \text{FeS}_2 + \text{O}_2 = \text{FeS} + \text{SO}_2 \)) or \( \text{H}_2\text{S} \) (via the reaction \( \text{FeS}_2 + \text{H}_2 = \text{FeS} + \text{H}_2\text{S} \)), depending on the redox conditions of metamorphism. If the former occurred, the \( \delta^{34}\text{S} \) value of pyrrhotite would decrease compared with preexisting pyrite due to liberation of high-\( \delta^{34}\text{S} \) \( \text{SO}_2 \). If the latter occurred, the \( \delta^{34}\text{S} \) value of pyrrhotite would increase compared with preexisting pyrite due to liberation of low-\( \delta^{34}\text{S} \) \( \text{H}_2\text{S} \) (Ohmoto, 1986). However, if no external source of MIF-bearing sulfur was involved in the formation of the Huronian sulfides, these

**FIG. 9.** Compiled multiple S isotope chemostratigraphy of the two studied drill cores in the Huronian Supergroup. All the sulfur four-isotope data were measured at WiscSIMS with improved methods (Ushikubo et al., 2014) from this study. The highlighted color bands are the range for \( \Delta^{33}\text{S} \) and \( \Delta^{36}\text{S} \). Analytical uncertainty is the measured 2SD reproducibility on the UWPy-1 pyrite standard. Red stars alongside the lithology column mark the stratigraphic position of analyzed samples in this study.
mass-dependent processes would neither create new MIF-S signatures nor destroy the original bulk MIF-S signatures. The dominant effect of regional metamorphism on $\Delta^{33}$S would be to homogenize and mute the variability of any preexisting MIF signals.

Although the Huronian sulfides may have experienced post depositional desulfidation reactions and homogenization in chemical composition, we regard that the measured $\Delta^{33}$S values may not deviate far from the average (whole rock) primary values. Given the rarity of very negative $\Delta^{33}$S values in pyrite at that time, to achieve a homogenized $\Delta^{33}$S near 0 value, it is unlikely that the $\Delta^{33}$S values of the primary Huronian sulfides can be significantly positive before homogenization by regional metamorphism. Supporting evidence for this view also comes from published bulk rock data of the McKim and Pecors formations (Wing et al., 2002, 2004). The bulk $\Delta^{33}$S values of these two formations ranges from $+0.22_{\infty}$ to $0_{\infty}$ which show similarly small and nonzero values compared with the SIMS data in this study.

5.3.2. Recycling of older MIF signals. It is also possible that fluids with preexisting MIF-S signals that are derived from recycled detrital pyrite may have overprinted the studied Huronian samples, resulting in the overall small MIF-S signals in the measured sections. A similar scenario has been used to explain the existence of small MIF-S signals during or shortly after the GOE when they were still not yet eliminated by chemical weathering and sediment dilution (Reinhard et al., 2013).

In this study, although only samples that were deposited in marine facies were investigated in an effort to avoid detrital pyrite (see Section 3.1), the clearest evidence—rounded detrital grains—would be destroyed by metamorphic recrystallization. It is thus possible that MIF-S signals...
Comparison of the WiscSIMS data and published data measured from the lower Huronian Supergroup

**Fig. 11.** A comparison between the data published in an early study (Papineau et al., 2007) and the new data set analyzed by WiscSIMS in this study. Highlighted color bands mark the data ranges of the lower Huronian Supergroup from this study. See the main text for detailed discussion.
from older sediments may have been remobilized. It is also possible that MIF signals can be mobilized by fluids from these detrital grains or from older rocks. For example, an independent study of S three-isotopes of detrital pyrite and pyrite overgrowths in the fluvial-deltaic Mississippian Formation of the Huronian Supergroup shows $\Delta^{33}S$ values ranging from $ca. -2^\%_{oo}$ to $ca. +1^\%_{oo}$ (Ulrich et al., 2011). These detrital pyrites can be a potential source of the measured MIF-$S$ signals in this study. More studies are suggested to further evaluate the contribution of recycled MIF-$S$ signals from late fluids in this region.

5.3.3. Dilution of MIF signals by magmatic fluids. Hypothetically, the overall small $\Delta^{33}S$ or $\Delta^{34}S$ signals can also result from the dilution of MIF signals by a magmatic reservoir with MDF signals. This process has been used to explain the general near-zero $\Delta^{33}S$ values in the Archean gold deposits in Western Australia (Xue et al., 2013). Notably, the Huronian Supergroup was intruded by multiple dikes (Fig. 1), including the Nipissing diabase at $ca. 2.22$ Ga (Fig. 2) (Cortus and Andrews, 1986; Mossman et al., 1993), which may have indeed caused a significant dilution of the primary MIF-$S$ signals. If true, potentially primary large MIF-$S$ signals of the Huronian Supergroup may have been significantly eroded by a large volume of MDF-bearing magmatic fluids.

Based on a mass balance perspective, if the primary MIF signal in the Huronian is pronounced, the MDF-bearing magmatic fluids should be very large in volume in order to achieve near-zero values. However, in light of the overall small $\Delta^{33}S$ or $\Delta^{34}S$ values analyzed from the remaining pyrite grains, the magmatic fluids that are needed to achieve overall near-zero values can be relatively small in volume. More studies are suggested to further evaluate the overall influence of late magmatic fluids in this region.

5.3.4. Small $\Delta^{33}S$ or $\Delta^{34}S$ signals produced by MDF. The small $\Delta^{33}S$ or $\Delta^{34}S$ signals in the Huronian can also be explained by the occurrence of MDF. Lab experiments of post-Archean samples show small magnitude of $\Delta^{34}S$ (from $0^\%_{oo}$ to $ca. 0.4^\%_{oo}$) or $\Delta^{34}S$ (from $-2.5^\%_{oo}$ to $0^\%_{oo}$) signals in sedimentary sulfides, which have been explained by the occurrence of MDF (Ono et al., 2006).

It was revealed that these small MDF-produced $\Delta^{33}S$ and $\Delta^{34}S$ signals have a distinctive $\Delta^{34}S/\Delta^{33}S$ slope of $-6.85$, which can be used as a tool to distinguish the origin of small $\Delta^{34}S$ and $\Delta^{33}S$ signals (Ono et al., 2006). However, due to the small range of values and larger uncertainty of data generated by SIMS compared with the conventional gas source mass spectrometer, it is not feasible to extract a reliable $\Delta^{34}S/\Delta^{33}S$ slope based on the current SIMS data set (Fig. 10). Therefore, conventional analysis of larger samples by gas source mass spectrometry may yield better precision and reveal subtle correlation even though the range of values is so small.

In summary, we propose a metamorphic origin for the Huronian pyrrhotite based on detailed investigation of petrography and mineral assemblages. Metamorphism tends to homogenize and mute variability but not create or destroy the MIF-$S$ signals. Recycled older MIF signals, dilution by magmatic fluids, or the occurrence of MDF may have also contributed to the generation of the overall small MIF-$S$ signals in the studied samples. Therefore, in contrast to the previous study on the same successions (Papineau et al., 2007), we argue that the precise location of the MIF–MDF transition in the Huronian remains unsolved. This putative MIF–MDF transition may have already been erased by postdepositional processes in the lower Huronian supergroup or may be located in the upper Huronian Supergroup.

5.4. Implications on the GOE

5.4.1. An evolving concept of the GOE. Although the existence of the GOE is indicated by sedimentological observations, the placement of the GOE in the Huronian stratigraphy is still ambiguous. Before discussing the GOE in the Huronian, it should be noted that the criteria to identify the GOE may vary among different authors.

Largely based on sedimentological observations, the GOE has been interpreted as a long time interval between $\sim 2.4$ and $\sim 2.0$ Ga, which is characterized by a gradual transition from reducing atmosphere to weakly oxygenated redox conditions in atmosphere and shallow oceans (Holland, 2002, 2006; Bekker, 2014; Gumsley et al., 2017). The discovery of the intimate response of time-series $S$ four-isotopes (i.e., MIF–MDF transition) to the rise of atmospheric oxygen makes it possible to pinpoint the timing of this putative event to be $ca. 2.45$ Ga (Farquhar et al., 2000; Farquhar and Wing, 2003, 2005). Thus, the GOE has been identified in two ways. In a broader sense, the GOE refers to a long-term sedimentological transition in the Paleoproterozoic with possible oscillations in atmospheric oxygen levels (Holland, 2006; Gumsley et al., 2017). In a narrower sense, the GOE is defined by the $S$ isotope transition from MIF to MDF. The latter has been increasingly adopted in studies of individual sections of the Paleoproterozoic (Guo et al., 2009; Williford et al., 2011; Luo et al., 2016).

The threshold $\Delta^{33}S$ value that marks the MIF–MDF transition and the GOE is uncertain. For example, chemostratigraphic studies from South Africa reveal a profound sulfur isotope transition from large MIF signals ($\Delta^{33}S$ up to $ca. +7^\%_{oo}$ $\Delta^{36}S$ down to $ca. -7^\%_{oo}$) and overall positive $\delta^{34}S$, to the onset of MDF with large MDF negative excursions in $\delta^{34}S$ (down to $ca. -30^\%_{oo}$) based on bulk $S$ isotope analysis (Fig. 12) (Cameron, 1982; Bekker et al., 2004; Luo et al., 2016). By defining the GOE as a transition to $|\Delta^{33}S|$ values $<0.5^\%_{oo}$ followed by a sharp negative $\delta^{34}S$ shift, the GOE was thereby pinpointed at $ca. 3$ m below the stratigraphic boundary between the Duitschland/Rooihoogte Formation and the Timeball Hill Formation in South Africa (Fig. 12) (Luo et al., 2016). Similarly, a detailed study of the Meteorite Bore Member of the Kungurra Formation (Turke Creek Group) from Western Australia reveals MIF signals with nonzero $\Delta^{33}S$ values ranging from $-0.83^\%_{oo}$ to $0.96^\%_{oo}$ (± 0.09%, 2SD), which have been interpreted to represent the final stage of the GOE (Williford et al., 2011). In this case, the threshold value of $|\Delta^{33}S| = ca. 0.9^\%_{oo}$ has been used to define the MIF–MDF transition and the GOE. Regardless of different threshold $|\Delta^{33}S|$ values in different studies, the approach of defining the GOE by MIF–MDF transition is practical in detailed chemostratigraphic studies of individual geological sections.

5.4.2. Differences of the $S$ isotope patterns of the GOE sections. The published and new $S$ isotope data show different features among the GOE sections in South Africa (Guo
et al., 2009; Luo et al., 2016; Gumsley et al., 2017), Western Australia (Williford et al., 2011), and North America (this study). We propose that the differences among these sections may result from the difference in analytical methods (bulk vs. SIMS) and the postdepositional history (early authigenic vs. late-stage metamorphism) of different basins. If the samples are well preserved, the integrated scanning electron microscopy (SEM) and secondary ion mass spectrometry (SIMS) techniques have the capability to distinguish and target small domains in different types of sulfide and to distinguish sulfur isotope signals that would otherwise be lost by bulk analysis. Therefore, a data set measured by SIMS provides more

FIG. 12. Multiple-S isotope profiles of Paleoproterozoic strata spanning the GOE in South Africa. Data in the y axis represent stratigraphic depth (in meters) of the drill cores. Source from the work of Luo et al. (2016). By defining the GOE as a transition to $\Delta^{33}S$ values $<0.5\%$, the GOE was pinpointed at ca. 3 m below the stratigraphic boundary between the Duitschland/Rooihoogte Formation and the Timeball Hill Formation in South Africa. For detailed information on stratigraphy, please refer to the work of Luo et al. (2016).
complexities compared with data analyzed from bulk samples (e.g., Williford et al., 2011).

In addition to analytical methods, various postdepositional history may also have contributed to the different patterns in S isotope profiles. In the studied Huronian Supergroup, low-grade metamorphism or late fluid activities may have influenced the S isotope values. Therefore, caution is needed when taking into account the different patterns in S isotope profiles. In the studied Huronian Supergroup, low-grade metamorphism or late fluid activities may have influenced the S isotope values. Therefore, caution is needed when conducting direct stratigraphic correlations among these sections (Hoffman, 2013; Gumsley et al., 2017).

The accuracy of geochronological data of the Paleoproterozoic successions is also critical to our understanding of the detailed dynamics of the GOE (Rasmussen et al., 2013; Gumsley et al., 2017). A recent compilation suggests that the GOE could potentially be lost from the record and reappear (Gumsley et al., 2017), either due to a fluctuating atmospheric composition (Gumsley et al., 2017) or due to geological reworking of older anomalous MIF-S sources (Reinhard et al., 2013). Therefore, the reconstruction of the GOE by chemostatigraphy of multiple S-isotopes should be based on an accurate geochronological framework.

6. Conclusions

(1) In this study, two drill cores of the Huronian Supergroup were revisited and analyzed for S four-isotopes in pyrite, chalcopyrite, and pyrrhotite. Based on the mineral assemblages and S isotopic compositions, it is proposed that the dominance of pyrrhotite in the studied drill cores results from regional metamorphism, instead of early authigenesis.

(2) The new $\Delta^{33}S$ and $\Delta^{34}S$ data in this study show homogeneous values at micrometer and millimeter scales. They all fall within a narrow range. The new $\Delta^{33}S$ data have improved precision compared to earlier studies. Small but analytically resolvable nonzero signals of $\Delta^{33}S$ (from $-0.07^{+0.04}_{-0.03}$ to $+0.38^{+0.02}_{-0.01}$) and $\Delta^{34}S$ (from $-4.1^{+0.3}_{-0.2}$ to $+1.0^{+0.4}_{-0.3}$) persist throughout the entire lower Huronian Supergroup. Neither pronounced MIF-S signals nor a MIF–MDF transition are seen in this study (Figs. 9 and 11).

(3) Four scenarios were discussed on the origin of small nonzero $\Delta^{33}S$ and $\Delta^{34}S$ signals: muted variability by regional metamorphism, recycling of older pyrite, dilution by magmatic fluids, and the genesis by MDF. Metamorphic recrystallization homogenized any isotope zoning in original sediments, and would have muted S isotope variability, but did not create or destroy the MIF-S signals. The involvement of recycled older MIF signals, magmatic fluids with MDF signals, or small nonzero signals produced by MDF may have also contributed to the overall muted signals. Therefore, we argue that the precise location of the MIF–MDF transition remains unsolved and may have been erased by post depositional processes in the lower Huronian Supergroup, or may be located in the upper Huronian Supergroup. The previously proposed MIF–MDF transition in the Huronian (Papineau et al., 2007) is not supported in this study and cannot be used to infer the GOE in North America.

(4) This study demonstrates that the SEM–SIMS technique provides fresh insight for unraveling the complexities of the GOE-bearing strata. The differences among Paleoproterozoic sections may result from differences in analytical methods (bulk vs. SIMS) and the postdepositional history (early authigenic vs. late-stage metamorphism) of different basins.

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Author Disclosure Statement

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Abbreviations Used

BSE = back-scattered electrons 
EDS = energy-dispersive spectrometry 
EPMA = electron-probe microanalysis 
GOE = Great Oxidation Event 
MDF = mass-dependent fractionation 
SEM = scanning electron microscope 
SIMS = secondary ion mass spectrometer 
VCDT = Vienna Canyon Diablo Troilite

Appendices can be found with online Supplementary Material.

Appendix 1: Time-series SIMS data and equations used for SIMS data correction.

Appendix 2: Integrated BSE, SE images and SIMS data for the Kerr–McGee drill core 150–4.

Appendix 3: Integrated BSE, SE images and SIMS data for the Kerr–McGee drill core 156–1.

Appendix 4: Data table of the S four-isotopes analyzed by SIMS.

Appendix 5: Data table of the elemental concentrations analyzed by EPMA.